





## ECONOMIC IN ECOLOGY

# Catalytic waste gas and exhaust air purification



## The system

EWK Environmental Engineering has been manufacturing systems for the purification of air and waste gas for decades. In the past few years the catalytic waste gas purification method has gained more and more importance in this context.

As a result of years of tests and developments we now use the SCR (selective catalytic reduction) method successfully for the reduction of pollutants in

process exhaust air

and

combustion waste gases.

The burning of **biological fuels**, (wood, straw, biogas etc.) as well as fossil fuels (heating oil, natural gas, heavy oil, coal) and industrial waste (solvent, sewage sludge, explosive, refuse and industrial residue etc.) generates large amounts of the following substances even in the most advanced furnaces:

- nitric oxide
- NOX CO • carbon monoxide
- hydrocarbon
- dioxin/furan
- C<sub>m</sub>H<sub>n</sub> PCDD/PCDF

With the SCR catalyzer technology these air pollutants can be precipitated to a large degree and modified to N<sub>2</sub>, CO<sub>2</sub> and H<sub>2</sub>O.



Catalytic waste gas purification for oil (heavy, extra light) or gas furnace



Catalytic waste gas purification for thermal waste utilization

$\mathrm{NO}_{\mathrm{X}}$ reaction with ammonia $\mathrm{NH}_3$	für NO für NO <sub>2</sub>	$\begin{array}{c} 4 \text{ NO} + 4 \text{ NH}_3 + \text{O}_2 \\ 6 \text{ NO}_2 + 8 \text{ NH}_3 + \text{O}_2 \end{array}$	$\rightarrow$	$\begin{array}{l} 4 \ {N_2} + 6 \ {H_2}{O} \\ 7 \ {N_2} + 12 \ {H_2}{O} + {O_2} \end{array}$
$NO_{\rm X}$ reaction with urea (NH <sub>2</sub> ) <sub>2</sub> CO	für NO für NO <sub>2</sub>	$\begin{array}{l} 4 \text{ NO} + 2 (\text{NH}_2)_2 \text{ CO} + 2 \text{ H}_2\text{O} + \text{O}_2 \\ 6 \text{ NO}_2 + 4 (\text{NH}_2)_2 \text{ CO} + 4 \text{ H}_2\text{O} \end{array}$	→ →	$\begin{array}{c} 4 \ N_2 + 6 \ H_2 O + 2 \ CO_2 \\ 7 \ N_2 + 12 \ H_2 O + 4 \ CO_2 \end{array}$
SO <sub>2</sub> secondary reaction		$2 SO_2 + O_2$ SO <sub>3</sub> + NH <sub>3</sub> + H <sub>2</sub> O SO <sub>3</sub> + 2 NH <sub>3</sub> + H <sub>2</sub> O	$\rightarrow$ $\rightarrow$ $\rightarrow$	2 SO <sub>3</sub> NH <sub>4</sub> HSO <sub>4</sub> (NH <sub>4</sub> ) <sub>2</sub> SO <sub>2</sub>
CO reaction		2 CO + O <sub>2</sub>	→	2 CO <sub>2</sub>
C <sub>m</sub> H <sub>n</sub> reaction		C <sub>m</sub> H <sub>n</sub> + O <sub>2</sub>	→	CO <sub>2</sub> +H <sub>2</sub> O



## The different methods

#### **Reduction catalyzers**

## NOx

The combustion in furnaces, motors and turbines generates large amounts of toxic nitrogen oxides ( $NO_X$ ), depending on temperature and fuel-bound nitrogen.

With the help of a reduction solution in the form of monatomic nitrogen compounds (urea, ammonia)  $NO_X$  can be converted (reduced) to harmless nitrogen and water vapor.

For that purpose an urea solution is sprayed into the crude gas flow and thermolyzed to NH<sub>3</sub>.

In the static mixer  $\tilde{\text{NO}}_{\text{X}}$  is thoroughly mixed with the  $\text{NH}_3.$ 

The reduction of  $NO_X$  to  $N_2$  and  $H_2O$  occurs in the DeNO<sub>X</sub> catalyzer. Honeycomb catalyzers made of homogenous ceramic base material with integrated active catalytic material are commonly used.

### **Oxidation catalyzers**

Flammable, partially toxic and odourintensive waste gases can be burned (oxidized) by thermal or catalytic methods. A suitable **catalyzer** allows an almost complete combustion **at significantly lower temperatures**. The catalyzer supports the chemical reaction of the gases **without additives** and without changing itself.

## Co/C<sub>m</sub>H<sub>n</sub>

Additional by-products of combustion or other industrial processes, carbon monoxides (CO) and hydrocarbons ( $C_mH_n$ ), are converted in the oxidation catalyzer.

In the process CO is oxidized to  $CO_2$  and the hydrocarbons burned to  $CO_2$  and  $H_2O$ .

Honeycomb catalyzers made of a ceramic base material with a coating of precious metal are commonly utilized.



Primarily in the burning of residual waste, but also in process waste gases, frequently present annular hydrocarbons can form dioxins and furans when they combine with chlorine. In an oxidation catalyzer, made of homogenous ceramic base material with integrated active catalytic material, the highly toxic pollutants are split up into harmless constituents.

#### Fiber filter catalyzer

To precipitate the micro dust (soot) of combustion engines fiber cartridge filters are used.

Due to the catalytic coating of the fibre the soot oxidizes at temperatures between 360 - 480 °C.







Reaction diagram of fiber filter catalyzer



DeNox catalyzer unit for gas-fired glass smelting process 130,700 Am<sup>3</sup>/h

## Data and facts

## Precipitation of catalytic systems for boiler and engine units

	Typical crude gas value	Obtainable clean gas value						
Fuel	N	Э <sub>х</sub>	С	0	Cm	H <sub>n</sub>	Dioxin	/Furan
	mg/	Nm³	mg/	Nm³	mg/	Nm <sup>3</sup>	ng/l	Nm³
Heating oil, heavy	800	80	40	-	20	-	-	-
Heating oil, extra light	400	50	15	-	10	-	-	-
Natural gas	250	30	10	-	100	30	-	-
Wood, waste, old timber	500	35	1000	100	300	50	0,3	<0,1
Residual waste incineration	200 - 4000	<100	1000 - 5000	<100	300 - 800	<50	8	<0,1
Diesel engine	2000 - 4000	<100	800	<100	50	<20	<20	<20
Natural gas engine	800	<100	1000	<100	<500	<50	<50	<50
Efficiency	90 -	98%	92 -	98%	65 -	90%	- 80	95%

#### **Temperature ranges**

Our catalyzers are marked by
low starting temperature and
wide temperature range with the middle value indicating the optimum operating level.

NO <sub>x</sub>	250 - <b>300</b> - 500 °C
Dioxin/Furan	220 - <b>300</b> - 420 °C
CO	180 - <b>260</b> - 600 °C
C <sub>m</sub> H <sub>n</sub>	120 - <b>x</b> - 600 °C
C	360 - <b>420</b> - 480 °C

**x**: For hydrocarbons the optimum operating temperature depends strongly on the pollutant composition.

## **Advantage**

- Urea is nontoxic, transportation and storage hold no risks
- NO<sub>X</sub> conversion rates of over 98% are possible
- No or neglectable reaction byproducts such as nitrous oxide, hydrocyanic acid, isocyanic acid
- All furnaces can be upgraded with SCR catalyzers
- Low MSR effort required
- Low ammonia slip

## **Applications**

- Thermal power station
- Waste incinerators
- Waste wood incineration
- Explosives incineration
- Gas/Diesel engines
- Gas turbines
- Crematories
- Solvent disposa
- Chemical/pharmaceutical industry
- Textile/paint/varnish industry
- Stainless steel pickling plants
- Various process air purifications
- Thermal residue utilization
- Cogeneration systems
- Greenhouses

#### Modular construction

- The reactor casing is constructed of stainless steel in a modular system with integrated insulation
- This allows an individual line-up of components depending on type and concentration of pollutant
- Single and multi-stage catalyzer layout as required
- Adjustments to structural conditions possible any time
- Compact and economical construction possible

## SCR catalyzer technology

- is characterized by
- Iong life span
- high quality standard
- simple operation
- high operating safety
- Iow maintenance effort
- low operating costs
- wide temperature range, 120-520°C, depending on pollutant
- high conversion rate of up to 98%
- low ammonia slip
- existing systems are easily upgraded

# Examples of SCR technology



Catalytic reactor with 6 rows of catalyzers for pickling plant



Catalytic convertor for  $NO_{\chi}$ , Dioxin, CO and HC behind incineration of explosives, 28,500 m<sup>3</sup>/h



2 Dual-DeNO<sub>X</sub> catalyzer units for heavy oil fired boiler, 56,000  $Am^3/h$  each



Dual-DeNO<sub>x</sub> catalyzer unit for glass smelters 304,600 m<sup>3</sup>/h



 $DeNO_X$  catalyzer unit for diesel engine 14,300 m<sup>3</sup>/h



Control panel with PLC system and dosing unit Top: electrical control unit with PLC Bottom: dosing unit for compressed air and reactant

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